Requested Patent:

EP0570838A1

Title:

S-TRIAZINE DERIVATIVES AS LIGHT STABILISERS.

Abstracted Patent:

EP0570838

Publication Date:

1993-11-24

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Application Number:

EP19930107781 19930513

Priority Number(s):

IT1992MI01203 19920519; US19930063747 19930520

IPC Classification:

C07D251/70; A61K7/44; C08K5/3492

Equivalents:

US5346691

ABSTRACT:

S-triazine derivatives of formula (I): a process for the preparation thereof and the use thereof as light stabilisers.

11) Publication number:

0 570 838 A1

(12)

EUROPEAN PATENT APPLICATION

(21) Application number: 93107781.2

(51) Int. Cl.5: C07D 251/70, A61K 7/44, C08K 5/3492

② Date of filing: 13.05.93

Priority: 19.05.92 IT MI921203

43 Date of publication of application: 24.11.93 Bulletin 93/47

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- 54 S-triazine derivatives as light stabilisers.
- 57 S-triazine derivatives of formula (I):

$$\begin{array}{c|c} & \text{NH-} & \text{-}\text{COR}_1 \\ & \text{NH-} & \text{-}\text{COR}_1 \\ & \text{NH-} & \text{-}\text{CO-}\text{X-R}_2 \\ \end{array}$$

a process for the preparation thereof and the use thereof as light stabilisers.

The present invention relates to s-triazine derivatives, the process for the preparation thereof and the use thereof as light stabilisers.

The compounds of the invention have the following formula (I):

$$\begin{array}{c} \text{NH-} \\ \text{N} \\ \text{NH-} \\ \text{NH-}$$

in which

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R is C₁-C₁₈ straight or branched alkyl, C₅-C₁₂ cycloalkyl optionally substituted with one or more C₁-C₄ alkyl;

X is oxygen or the -NH- group;

 R_1 has the same meanings as R or it is hydrogen, an alkali metal, an ammonium group or a group of formula (II):

$$A = \begin{bmatrix} O - CH_2 & CH & - \\ R_3 & - \\ \end{bmatrix}_n$$
 (II)

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in which A is C_1 - C_8 straight or branched alkyl, C_5 - C_8 cycloalkyl, aryl optionally substituted with one or more C_1 - C_4 alkyl, R_3 is hydrogen or methyl and n can be an integer from 1 to 10;

R₂ has the same meanings as R when X is the -NH- group;

and it has the same meanings as R₁ when X is oxygen.

Examples of C₁-C₁₈ alkyl groups comprise methyl, ethyl, isopropyl, t-butyl, n-hexyl, 2-ethylbutyl, 2-ethylhexyl, 1,1,3,3-tetramethylbutyl.

Examples of C_5 - C_8 or C_5 - C_{12} cycloalkyl groups optionally substituted with C_1 - C_4 alkyl comprise cyclohexyl, 2-, 3- or 4-methylcyclohexyl, cyclopentyl.

Examples of aryl groups optionally substituted with C_1 - C_4 alkyl comprise phenyl, 2-, 3- or 4-methylphenyl.

Preferred compounds of formula (I) are those in which X is oxygen and R_1 and R_2 , which are the same, are C_1 - C_{18} straight or branched alkyl groups, C_5 - C_{12} cycloalkyl groups optionally substituted with C_1 - C_4 alkyl or a group of formula (II) wherein R_3 is hydrogen, A is C_1 - C_{18} alkyl and n is 2.

Preferred compounds of formula (I) also are those in which X is -NH- and R_1 and R_2 , which are the same or different, are C_1 - C_{18} straight or branched alkyl groups or C_5 - C_{12} cycloalkyl groups optionally substituted with C_1 - C_4 alkyl groups.

Ultraviolet radiations of sunlight are known to exert a damaging action on skin tissue and to cause the degradation of polymers. By using particular compounds, the so-called sunscreens, which are capable of absorbing the UV part of solar radiation, the damaging effects and the aging of the skin and polymer materials can be prevented or, at least, slowed down.

A number of substances have been studied and tested as protecting agents, and an extensive patent literature exists on this subject, in which compounds belonging to different chemical classes are proposed, which are capable of absorbing in the ultraviolet region, particularly the radiation from 290 to 320 nm, the so-called UV-B, that is the most noxious.

Only a few of the compounds proposed up to now as sun screens proved suitable for the practical application. Among these, p-methoxy-cinnamic acid and p-dimethylaminobenzoic acid esters, benzotriazoles and hydroxybenzophenones.

A common drawback of all these compounds is the low power thereof to absorb the radiation from 290 to 320 nm, which makes necessary to use relatively large amounts thereof in cosmetic compositions to obtain an optimum light-protecting capability.

An optimum UV absorber should have the following characteristics:

- 1) high specific extinction E' at 290-320 nm, thus resulting in a low dosage and accordingly cost-savings and a minimum toxicological risk;
- 2) light stability;

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- 3) heat stability;
- 4) oxidation stability;
- 5) stability at different pH;
 - 6) good solubility in base substances commonly used for the preparation of cosmetic formulations;
 - 7) negligible toxicity;
 - 8) colour and odour which are compatible with the envisaged uses;
 - 9) comparatively high molecular weight, therefore with a lower probability of absorption by the skin and a higher safety from the toxicological point of view;
 - 10) compatibility with the various substances generally used in dermatological formulations.

In DE 3 205 398 Patent, s-triazine derivatives are disclosed, obtained by reacting trichlorotriazine with p-aminobenzoic acid esters, which derivatives highly absorb at the UV-B region. However, these compounds are very poorly soluble in the solvents commonly used in the formulation of sun creams, thus making them very difficult to use, particularly when an increased percentage of the light-protecting agent is needed to prepare formulations having a high sun protection factor.

The sun protection factor (SPF) is a unit of measurement of the light-protecting power of a sunscreen or of a cosmetic formulation containing one or more sunscreens. It is directly related to the specific extinction and accordingly to the light-protecting agent present in the cosmetic formulation.

Now it has been found that the compounds according to present invention, besides absorbing very highly in the UV-B region and having a very high SPF, surprisingly show an excellent solubility in the solvents commonly used as components in the sunscreens formulations.

Therefore, a further object of the invention resides in the use of the compounds of formula (I) as sun screens and light stabilisers, thanks to the capability thereof of exerting a surprising protecting activity on the skin from the noxious components of sunlight radiation.

The compounds of the invention are also valuable for use in light stabilization of synthetic polymers, in order to prevent light degradations and alterations.

Compounds (I), besides having a very high absorption in the UV-B region as well as an excellent solubility, have other characteristics necessary for the application in practice, such as heat stability, no toxicity and the like.

The compounds of the present invention can be prepared by reacting triazine derivatives of formula (III)

$$R-NH-\infty-$$

$$N-NH-\infty$$

with p-aminobenzoic acid esters of formula (IV)

$$H_2N-COOR_1$$

or by reacting triazine derivatives of formula (V)

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$$R_{1} \infty C - \left(\begin{array}{c} N \\ \\ N \end{array} \right) - NH - \left(\begin{array}{c} N \\ \\ N \end{array} \right)$$

$$(V)$$

with p-aminobenzoic acid amides of formula (VI)

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In formulae (III)-(VI), z is bromine or preferably chlorine whereas R and R₁ have the meanings described above.

The compounds of formulae (III)-(VI) are known or they can be prepared according to known methods. The reaction is carried out at a temperature from 50 to 200 °C in a suitable solvent.

Suitable solvents in which the reaction is carried out are, for example, acetonitrile; ketones, such as acetone, methyl ethyl ketone; ethers such as diethyl ether, diisopropyl ether, tetrahydrofuran, dioxane; aliphatic or aromatic hydrocarbons such as pentane, heptane, cyclohexane, benzene, toluene, xylene or mixtures thereof; aliphatic carboxylic acid esters such as ethyl acetate.

The reaction can be effected either in the absence or in the presence of acid acceptors, such as alkali or alkaline-earth metal hydroxides, alkali metal bicarbonates or carbonates, in molar ratios of 2-3 moles of compounds of formula (IV) or (VI) to mole of compounds of formula (III) or (V).

The intermediates of formula (III) and (V), before the subsequent reaction with the compounds of formula (IV) and (VI), can be recovered and purified or, more easily, they are reacted as crude products, considering them as first step of a multi-step synthesis.

The reactions of trihalotriazines to replace the three halogen atoms with amino residues, which can be the same or different, are known and widely described in technical literature, particularly in the literature concerning some kinds of dyes and optical bleachers.

According to one of the preferred embodiments of the invention, the compositions containing the compounds of formula (I) are used to protect the skin from the damaging effects of sunlight radiations.

The compounds according to the present invention can be added, of course also in combination with other stabilizers, to the cosmetic formulations as well as to synthetic polymers, generally in amounts ranging from 0.05 to 15%, preferably from 0.1 to 10% by weight of the polymer or cosmetic formulation.

The cosmetic formulations can be of various kinds and they can be used for different purposes. Generally they are in form of ointments, creams, lotions, emulsions.

The compounds of formula (I) are added either to protect the formulations themselves, for example to prevent undesired discolourations, or to protect the skin treated with the formulation from the damaging action of UV-B radiations, which causes erythema and accelerates the aging of the skin making it prematurely dry, wrinkled or squamous.

The following examples illustrate the invention.

Example 1

A mixture of 167 g of p-nitrobenzoic acid, 204 g of 2-ethyl-1-butanol and 3.5 g of p-toluenesulfonic acid is refluxed and stirred for 2 hours, separating the formed water. After that, the alcohol excess is distilled off, then the formed product is distilled at 150 °C and 0.04 mmHg, to obtain 209 g of 2-ethylbutyl p-nitrobenzoate in form of a light yellow viscous liquid.

Examples 2-8

Following the procedure described in Example 1, recovering and purifying the products by distillation or crystallization, the esters listed in Table 1 are prepared.

TABLE 1

0₂N - 0 - COOR

Ex	ample	R 1	M.P. °C or E.P.
⁵ 2	(0	CH ₃) ₂ CH-CH ₂ -	66-68
3	сн	(0-CH ₂ -CH ₂) ₂	not det.
4		CH3 -	not det.
25 !	5 сн	3 CH3 CH3	78-80
30 (c c	14 ^H 29	50-52
1		4 ^H 9 ^{-CH(C} 2 ^H 3-CH ₂	not det.
35 (C	H ₃	158-160/0.5 mbar

Example 9

32.5 g of isopropylamine dissolved in 250 ml of acetone are added with 71 g of a 30% sodium hydroxide aqueous solution. This mixture, cooled to 0°C, is slowly added with 92.7 g of p-nitrobenzoyl chloride dissolved in 300 ml of acetone, cooling with an ice bath to keep temperature from 0 to 5°C. Subsequently stirring is continued for 1 more hour allowing temperature to raise to 20-30°C. Then the reaction mixture is poured into cold water. The formed precipitate is filtered, washed with water, dried and crystallized from isopropanol. 77 g of N-isopropyl-p-nitrobenzamide are obtained as a whitish solid substance with m.p. 152-154°C

Examples 10-15

Following the procedure of Example 9, the p-nitrobenzamides listed in Table 2 are obtained.

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TABLE 2

02 N- 0 -CO-NH-

10	Example	R	M.P. (°C)
10	10	(CH3)3C-	159-161
15	11	CH ₃ -	212-214
	12	n-C ₆ H ₁₃ -	86-88
20	13	C4H9-CH(C2H5)-CH2-	not det.
,	14	(CH ₃) ₃ C-CH ₂ -C(CH ₃) ₂	114-116
25	15	<u></u>	202-204

Example 16

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200 g of 1-ethylbutyl p-nitrobenzoate of Example 1, 900 ml of methanol, 12 g of acetic acid and 16 g of 50% Raney Nickel in water are placed into an autoclave. The autoclave is washed with nitrogen, then with hydrogen. After that, the nitro derivative is reduced at a temperature of 75 °C and under a pressure of 30 bar. The catalyst is filtered off, then the mixture is evaporated to dryness and the residual product is crystallized with hexane.

132 g of 2-ethylbutyl p-aminobenzoate are obtained, as a whitish substance with m.p. 52-53 °C.

40 Examples 17-29

Following the procedure of Example 16, by hydrogenating the compounds of Examples 2-15, the compounds listed in Table 3 are obtained.

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TABLE 3

Example	R	x	M.P. (°C) or E.P.
17	(сн ₃) сн-	нн	151-153°C
18	сн ₃ —	0	109-111
19	(CH ₃) ₂ CH-CH ₂ -	0	57-59
20	(CH ₃) ₃ C -	NH	124-126
21	сн ₃ —	NH	177-179
22	CH-(0-CH2-CH2)2-	0	204-206/1mbar
23	n-C ₆ H ₁₃ -	нн	107-109
24	C4H9-CH(C2H5)-CH2-	0	49-51
25	C4H9-CH(C2H5)-CH2	NH	110-112
26 .	CH3	0	62-63
27	CH3/	NH	181 - 183

55 Example 30

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A solution of 37 g of trichlorotriazine in 450 ml of acetone, cooled to 0 °C, is added with 17.6 g of sodium bicarbonate, then slowly with 39 g of tert-butyl p-aminobenzamide (compound of Example 20),

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63-65

188-190/0.5mbar

keeping temperature at 0 °C by cooling. Subsequently the mixture is stirred for 30 minutes, then 150 ml of water are added and stirring is continued for 1/2 hour; then the precipitate is filtered, washed many times with water and dried under vacuum.

64 g of dichlorotriazine derivative are obtained as a white substance with m.p. > 250 °C.

Examples 31-39

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Following the procedure in Example 30, starting from trichlorotriazine and compounds of Examples 16-29, the dichlorotriazine derivatives listed in Table 4 are prepared.

TABLE 4

Exam	ple R ₂	Х	M.P.(°C)
31	CH3-	NH	>260
32	CH3-	0	>260
33	(CH ₃) ₂ CH-	NH	>260
34	C4H9-CH(C2H5)-CH2	0	245-248
35	C4H9-CH(C2H5)-CH2	- мн	250-252
36	(CH ₃) ₃ C-CH ₂ -C(CH	1 ₃) ₂ - NH	240-243
37	nC ₆ H ₁ -	NH	220-223
38	C ₁₄ H ₂₉	0	205-207
39	<u> </u>	NH	>260

Example 40

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17 g of the compound of Example 30 and 26 g 2-ethylhexyl p-aminobenzoate (Example 24) in 250 ml of xylene are stirred under reflux for 4 hours. Then xylene is distilled off and the residue is recrystallized from a toluene and hexane mixture.

35 g of compound of formula (I) are obtained (R = $(CH_3)_3C_7$, R₁ = R₂ = C₄ H₉-CH(C₂H₅)-CH₂-; X = 0) as a white substance with m.p. 101-103 °C.

Examples 41-53

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Following the procedure of Example 40, reacting the triazine derivatives of Table 4 with the compounds of Table 3, the compounds of formula (I) listed in Table 5 are prepared.

10	1. 1 1 2 1 2 2 2 2 2 2 2 2 2 2 2 2 2 2 2 2	M.P. (•C)	175- 178	108-110	149-152	91-93	148-151	129-132	continued -
15	 	1 1 1 1	0	0	•	•		0	ı
26		R.S.	P. P	C449CH(C2H5)-CH2	, , , , , , , , , , , , , , , , , , ,	C4H2CH(C2H5)CH2+	C2H5-CH(C2H5)-CH2-	e V	•
30 35	TABLE 5	ec .	# _e & & & & & & & & & & & & & & & & & & &	C449-CH(C2H2)CH2-	£	C4H9-CH(C2H2)-CH2- C4H9CH(C2H2)CH2+	s-ch(c ₂ h ₅)-ch ₂	£",	
40									
4 5		e c	- E	Сн.	(CH ₃) ₃ C.	CH ₃) ₃ C-	\bigcirc	n-C ₆ H ₁₃ -	
50		Example	.	45	43	4	\$	94	

Example	le na		R ₂		M.P. (•C)
47	C4H9-CH(C2H2)-CH2-	CH ₃ -	C4H9CH(C2H5)-CH2-	Ŧ.	231-233
4	C4H9-CH(C2H5)-CH2	C4H9-CH(C2H3)-CH2- C4H9CH(C2H3)-CH2-	C4HgCH(C2H5)-CH2-	0	91-93
6	(CH ₃) ₃ C-CH ₂ -0(CH ₃) ₂ -	(CH ₃) ₂ -CH-CH ₂ -	(CH ₃) ₂ -CH-CH ₂ -	0	152-154
8	C4H9CH(C2H3-CH2-	C4H9-CH(C2H5)-CH2 C4H9-CH(C2H5)-CH2	С49-СН(С2Н5)-СН2	H	104-106
15	(CH ₃) ₃ C-CH ₂ -C(CH ₃) ₋	сн ³ (-0-сн ² сн ^{2) 2-}	CH3 (-0-CH2-CH2)2-	0	69-71
25	(CH ₃) ₂ CH-	C14H29	(CH ₃) ₂ -CH-	Ę	109-112
53	(CH ₃) ₂ -CH-	- CH3-CH(CH2)-CH2-CH(CH2)-CH2-	C4H9-CH(C2H5)-CH2-	0	110-112

In the following Table 6, the solubilities in isopropyl myristate (which is the solvent generally used for the preparation of sun creams) and the specific extinctions of the compounds of formula (I) are compared with those of the product of DE 3 205 398 patent (compound C), which is commercially available.

TABLE 6

Example	max (nm)	E'	Solubility in isopropyl myristate at 20 °C g/100 ml
42	311	1647	> 20
43	311	1624	> 20
45	310	1613	> 20
46	311	1556	> 20
49	311	1689	> 20
50	310	1449	> 20
53	310	1607	> 20
Compound C	313	1563	< 5

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Example 54

Preparation of a sun cream

Mixture A

A mixture of 44 g of water, 0.5 g of the thickening agent Synthalen L (cross-linked polyacrylic acid), 5 g of glycerin, 0,3 g of imidazolidinylurea and 0.15 g of methyl p-hydroxybenzoate is prepared.

Mixture B

5 g of the compound of Example 43 are dissolved in 25 g of isopropyl myristate, the solution is added with 5 g of vaseline oil, 2 g of cetyl alcohol, 5 g of glycerin monostearate, 5 g of capric/caprylic triglyceride, 2 g of polysorbate 20, 0,2 g of bee wax and 0.05 g of propyl p-hydroxybenzoate.

The mixtures A and B are heated separately to 70 °C, then mixture B is added to mixture A with stirring. The resulting emulsion is neutralized adding 0.75 g of triethanolamine and finally 0.02 g of perfume are added at a temperature below 35 °C.

Claims

Claims for the following Contracting States: AT, BE, CH, LI, DE, ES, FR, GB, NL

1. Compounds of general formula (I):

40 $NH - COOR_{1}$ 45 $R - NH - CO - NH - NH - CO - X - R_{2}$ (1)

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in which

R is C_1 - C_{18} straight or branched alkyl, C_5 - C_{12} cycloalkyl optionally substituted with one or more C_1 - C_4 alkyl;

X is oxygen or the -NH- group;

 R_1 has the same meanings as R or it is hydrogen, an alkali metal, an ammonium group or a group of formula (II):

$$A = \begin{bmatrix} O - CH_2 - CH - \\ R_3 \end{bmatrix}_n$$
 (II)

in which A is C_1 - C_8 straight or branched alkyl, C_5 - C_8 cycloalkyl, aryl optionally substituted with one or more C_1 - C_4 alkyl, R_3 is hydrogen or methyl and n can be an integer from 1 to 10; R_2 has the same meanings as R when X is the -NH- group; and it has the same meanings as R_1 when X is oxygen.

- 2. Compounds according to claim 1 in which X is oxygen and R₁ and R₂, which are the same, are C₁-C₁₈ straight or branched alkyl groups, C₅-C₁₂ cycloalkyl groups optionally substituted with C₁-C₄ alkyl or a group of formula (II) wherein R₃ is hydrogen, A is C₁-C₁₈ alkyl and n is 2.
- 3. Compounds according to claim 1 in which X is -NH-and R₁ and R₂, which are the same or different, are C₁-C₁₈ straight or branched alkyl groups or C₅-C₁₂ cycloalkyl groups optionally substituted with C₁-C₄ alkyl groups.
 - 4. A process for the preparation of the compounds according to claim 1, which process comprises reacting a s-triazine derivative of formula (III) or (V):

$$\begin{array}{c|c}
R-NH-CO-\\
\end{array}$$

$$\begin{array}{c|c}
\end{array}$$

$$\begin{array}{c|c}$$

$$\begin{array}{c|c}
\end{array}$$

$$\begin{array}{c|c}
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respectively with compounds of formula (IV) or (VI):

$$H_2N-COOR_1$$
 $R-NH-CO-COID-NH_2$

- in which, in formulae (III) and (V), Z is bromine or chlorine, R, R₁ have the meanings defined above, at a temperature from 50 to 200 °C in a suitable solvent.
- 5. The use of the compounds of claims 1-3, as light stabilisers for synthetic polymers.

6. The use of the compounds of claims 1-3, as agents useful in dermatology.

- 7. The use of the compounds of claims 1-3, as sunscreens for the preparation of cosmetic and dermatological compositions useful for the protection of the skin.
- 8. Cosmetic and dermatological compositions containing one or more compounds of claims 1-3.
- 9. Cosmetic and dermatological compositions according to claim 8 containing from 0.05 to 10% by weight of the compounds of formula (I).



EUROPEAN SEARCH REPORT

Application Number

EP 93 10 7781

ategory	Citation of document with in of relevant pas	dication, where appropriate,	Relevant to claim	CLASSIFICATION OF THE APPLICATION (Int. CL5)
	EP-A-0 087 098 (BASF * complete document	AG)	1,5,6	C07D251/70 A61K7/44 C08K5/3492
				TECHNICAL FIELDS SEARCHED (Int. CL5) CO7D A61K
	The present search report has l	neen drawn up for all claims	arth	Examinor
	THE HAGUE	27 JULY 1993		VAN BIJLEN H.
A:t	CATEGORY OF CITED DOCUME articularly relevant if taken alone articularly relevant if combined with ar ocument of the same category echnological background one-written disclosure ntermediate document	ENTS T: theory or E: earlier p refer the other D: documen L: documen	r principle underlying atent document, but p filing date at cited in the applicat it cited for other reaso of the same patent fa	ublishes on, or tion ns